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Correct Answer: C. The final temperature of the lead-water system will be lower than the final temperature of the copper-water system. A blacksmith heats a 1,540 g iron horseshoe to a temperature of 1445°C before dropping it into 4,280 g of water at 23.1°C.

Student Exploration Calorimetry Lab

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Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water. A dynamic graph (temperature vs. time) shows temperatures of the individual substances after mixing.

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Vocabulary: calorie, calorimeter, joule, specific heat capacity
Prior Knowledge Questions (Do these BEFORE using the Gizmo.)
1. The Latin word calor means "heat," and meter comes from the Greek word meaning "to measure." What do you think a calorimeter does? Measures heat
2. Where have you heard the word calorie before? ? What do you think a calorie

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mixing. Calorimetry Lab Gizmo :
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A calorimeter is an insulated container filled with a liquid, usually water. When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up. Calorimeters can be used to find a
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Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. Modify initial mass and temperature values to see effects on the system. One or any combination of the substances can be mixed with water. A dynamic graph (temperature vs. time) shows temperatures of the individual substances after mixing.

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investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. ****YOU WILL NEED YOUR OWN GIZMO ACCOUNT TO COMPLETE THIS ACTIVITY. INSTRUCTIONS ARE HERE** (be sure to record your login info on Pg G of your notebook)

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